

MONO LAKE – POSSIBLE FUTURES

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EXTENDED ABSTRACT

Mono Lake is a hypersaline lake, having approximately 7.3 % salt content, and consisting mainly of sodium salts of chlorides, carbonates, and sulfates. The lake is situated in eastern California, USA, and is fed by five mountain streams. The lake has an area of ~ 182 km², a volume of ~ 3.22 km³, and a maximum depth of 48.8 m. The lake had a salinity of 80 gm(s)/L in the year 2002. Additionally Mono Lake is ~ 760,000 years old, one of the oldest in North America. In 1994 in the attempt to reverse the desiccation of the lake, the California State Water Resources Board issued D 1631. It concerned the flows and maintenance of the future of the lake. [1], [2], [3], [4]

This paper considers three possible futures:

- I. A Salinity Source For PRO, Pressure Retarded Osmosis
- II. Developing A Small Fresh Water Lake
- III. A Lake Complying With The Water Resources Board

In Number I, several forms of the van't Hoff equation are compared for the pressure P, in atmospheres, using the Mono Lake salinity, and the "sea water" salinity.

In Number II, equations for the mass m, gm(t)*af/L, the volume V, af, and salt content Z, gm(s)*af/L are developed. Then, for assumed initial conditions, solutions are given in a table and a graph.

In Number III, Initial flow conditions are given that will reverse the desiccation and satisfy the 20 year Water Resources Board plan.

M = mass, gm(t)*af/L, V = volume, af,
Z = salt content, gm(s)*af/L, G = salinity, gm(s),
R = density, gm(t)/L, S = % salt, gm(s)/100 gm(t).

I. A Salinity Power Source for PRO, Pressure Retarded Osmosis

A comparison is developed, in atmospheres, between the pressure developed from the salinity of "sea water", and the pressure developed from the salinity of Mono Lake.

Several forms of the van't Hoff equation are used.[5], [6], [7]

Osmotic Pressure

$$\Delta P = P_s - P_f \quad (1)$$

Where

$$P = MR'T = (G/M_w)*R'T \quad (2)$$

And so

$$P_f = (G_f/M_w)*R'T \quad (3)$$

For fresh water, see Section IIb,

$$G_f = 0.064 \text{ gm(s)/L}$$

$$M_w \sim 80 \text{ gm(s)/mol}$$

So

$$P_f = (0.064/80)*24.618 = 0.01969 \text{ atm} \sim 0 \quad (4)$$

Thus

$$\Delta P = P_s - P = P \quad (5)$$

In the above

$$\begin{aligned} \Delta P &= \text{osmotic Pressure, atm} & R' &= 0.08206 \text{ L} \cdot \text{atm/mol} \cdot \text{K} \\ P_s &= P \text{ salt solution, atm} & T &= 300 \text{ K} \\ P_f &= P \text{ fresh water, atm} & RT &= 24.618 \text{ L} \cdot \text{atm/mol} \\ M &= \text{mol/L, molarity} & M_w &= \text{gm(s)/mol} \\ P &= \text{Pressure, atm} & G &= \text{gm(s)/L} \end{aligned}$$

Now from references [8] and [9], for “sea water” ~ 85% NaCl,

Table I									
P, bar	G	M	a	i	Mw	H	kwh/m ³		
27	--	--	--	--	270	0.7			
20.3	35	0.6	0.676	2	58.5		203	0.563	

Where

$$\begin{aligned} i &= \text{ions per molecule, dimensionless} \\ a &= \text{coefficient of activity, dimensionless} \\ H &= \text{column height of water, m} \\ 1 \text{ atm} &= 1.0135 \cdot \text{bar} \end{aligned}$$

Now consider

$$P = MR'T \quad (6)$$

$$P = iaMR'T \quad (7)$$

$$P = iMR'T = RT' \cdot \text{Sum } M_j \quad (8)$$

Thus from [8]

$$P = MR'T = (0.6) \cdot 24.618 = 14.7708 \text{ atm} \quad (6a)$$

$$P = iaMR'T = 2 \cdot (0.676) \cdot (0.6) \cdot 24.618 = 19.970 \text{ atm} \quad (7a)$$

$$P = iMR'T = 2 \cdot (0.6) \cdot 24.618 = 29.5416 \text{ atm} \quad (8a)$$

From [7]

$$P = 26.66 \text{ atm} \quad (8b)$$

From the “sea water” analysis, we can see that equations (6a) and (8a) provide the approximate lower and upper marginal boundaries, for the estimation of the pressure P, in atmospheres.

I. A Salinity Source For PRO, Pressure Retarded Osmosis

Now from references [9] and [14]

Table II					
m	Sum m _j	G	Rho	T(assumed)	
1.09	2.06	90	1071.5	300	

Now

$$M = \text{Rho} \cdot m / 1000 = \text{gm(t)/L} \cdot (\text{mol/kg(t)}) \cdot (\text{kg(t)/gm(t)}) = \text{mol/L} \quad (9)$$

And

$$\text{Sum } m_j = m_t, \quad (10)$$

$$M_t = \text{Rho} \cdot m_t / 1000 = \text{Sum } M_j = \text{Rho} \cdot \text{Sum } m_j / 1000 \quad (11)$$

Where

$$\begin{aligned} m &= \text{mol/1000 gm(t)} & R' &= 0.08206 \text{ L atm/mol K} \\ \text{Rho} &= \text{gm(t)/L} & T &= 300 \text{ K} \\ M &= \text{mol/L} & R'T &= 24.618 \text{ L atm/mol} \end{aligned}$$

Thus

$$M = \text{Rho} \cdot m / 1000 = 1071.5 \cdot 1.09 / 1000 = 1.16795 \text{ mol/L} \quad (12)$$

$$\text{Sum } M = \text{Rho} \cdot \text{Sum } m_j = 1075.1 \cdot 2.06 / 1000 = 2.2073 \text{ mol/L} \quad (13)$$

Thus

$$P = MR'T = 1.16795 \cdot 24.618 = 28.752 \text{ atm} \quad (6b)$$

$$P = R'T \cdot \sum M_j = 24.618 \cdot 2.2073 = 54.339 \text{ atm} \quad (8b)$$

Now

$$Ch = 10.333 \text{ m/atm}$$

$$Cw = 0.028146 \text{ kwh/m}^3 \text{ atm}$$

So

$$H = Ch \cdot P, \text{ m} \quad (14)$$

$$\text{kwh/m}^3 = Cw \cdot P \quad (15)$$

$$\text{kwh/af} = \text{kwh/m}^3 \cdot \text{m}^3/\text{af} = 1233.47 \cdot Cw \cdot P \quad (16)$$

$$\text{kwh/y} = \text{kwh/af} \cdot \text{af/y} = 1223.47 \cdot Cw \cdot P \cdot V' \quad (17)$$

$$\text{kw} = \text{kwh/y} \cdot \text{y/h} = 1233.47 \cdot Cw \cdot P \cdot V' / 8760 \quad (18)$$

Thus for $P = 28.752 \text{ atm}$, the lower boundary for Mono Lake, with $V' = 10^5 \text{ af/y}$, results in

$$H = 297.094 \text{ m} \quad (14a)$$

$$\text{kwh/m}^3 = 0.80925 \text{ kwh/m}^3 \quad (15a)$$

$$\text{kwh/af} = 998.190 \text{ kwh/af} \quad (16a)$$

$$\text{kwh/y} = 99.819 \cdot 10^6 \text{ kwh/y} \quad (17a)$$

$$\text{kw} = 11,394 \text{ kw} \quad (18a)$$

For $P = 54.339$, the upper boundary for Mono Lake, and $V' = 10^5 \text{ af/y}$, results in

$$H = 561.484 \text{ m} \quad (14b)$$

$$\text{kwh/m}^3 = 1.5294 \text{ kwh/m}^3 \quad (15b)$$

$$\text{kwh/af} = 1866.50 \text{ kwh/af} \quad (16b)$$

$$\text{kwh/y} = 188.65 \cdot 10^6 \text{ kwh/y} \quad (17b)$$

$$\text{kw} = 21,535 \text{ kw} \quad (18b)$$

These results provide the approximate lower and upper bounds, when $G = 90 \text{ gm(s)/L}$, for Mono Lake.

I. A Salinity Source For PRO, Pressure Retarded Osmosis

Now Table III contains, for the lower boundary, some necessary conditions, in order to extend the analysis.

Table III		
P, atm	G, gm(s)/L	R'T, L atm/mol K
28.752	90	24.618

Now

$$P = MR'T = (G/Mw) \cdot R'T \quad (19)$$

Thus

$$\begin{aligned} Mw &= G \cdot R'T / P = 90 \cdot 24.618 / 28.752 = 77.0597 \text{ gm(s)/mol} \\ &= \text{mixture molecular weight} \end{aligned} \quad (20)$$

So

$$P = G \cdot R'T / Mw = G \cdot 24.618 / 77.0597 = 0.31946 \cdot G \quad (21)$$

$$P(G) = G / 3.1302, \text{ atm} \quad (22)$$

In reference [14], for Mono Lake

$$36 < G < 698, \text{ gm(s)/L}$$

Thus, if $G = 313.02 \text{ gm(s)/L}$, then

$$P = 100 \text{ atm} \quad (23)$$

$$H = 1033.3 \text{ m} \quad (24)$$

$$\text{kwh/m}^3 = 2.8146 \text{ kwh/m}^3 \quad (25)$$

These results indicate the future potential of Mono Lake. The characteristics values of Mono Lake exceed the characteristic values of "sea water", for developing osmotic

power, PRO. This should result, in serious consideration, for a Mono Lake PRO site, in the future.

Ila. Developing A Small Fresh Water Lake

Controlling Equations [10]

Mass, $M(t)$

$$M(t) = R(t)*V(t) = \text{Sum } M_j(t) \quad (1a)$$

$$M'(t) = R(t)*V'(t) + R'(t)*V(t) \quad (1b)$$

Where

$$R(t) = \text{gm}(t)/L, \text{ density}$$

$$V(t) = \text{af, volume}$$

$$M(t) = \text{gm}(t)*\text{af}/L$$

Also

$$M'(t) = \text{Sum } M_j'(t) = M_{in}'(t) + M_{ra}'(t) + M_{sp}'(t) + M_{ev}'(t) + M_{se}'(t) + M_{ex}'(t) \quad (1c)$$

Where

$$M_{in}'(t) = R_{in} * V_{in}'(t)$$

$$M_{ra}'(t) = -R_o * V_{ra}'(t)$$

$$M_{sp}'(t) = R_{sp} * V_{sp}'(t)$$

$$M_{ev}'(t) = -R_o * V_{ev}'(t)$$

$$M_{se}'(t) = -R(t) * V_{se}'(t)$$

$$M_{ex}'(t) = -R(t) * V_{ex}'(t)$$

Thus

$$\begin{aligned} M'(t) &= R(t)*V'(t) + R'(t)*V(t) \\ &= R_{in} * V_{in}'(t) + R_o * V_{ra}'(t) + R_{sp} * V_{sp}'(t) - R_o * V_{ev}'(t) \\ &\quad - R(t) * [V_{se}'(t) + V_{ex}'(t)] \end{aligned} \quad (2)$$

Volume, $V(t)$

$$\begin{aligned} V'(t) &= \{R_{in} * V_{in}'(t) + R_o * V_{ra}'(t) + R_{sp} * V_{sp}'(t) - R_o * V_{ev}'(t) - R'(t) * V(t)\} / R(t) \\ &\quad - V_{sp}'(t) - V_{ex}'(t), \quad V(0) = V_o \end{aligned} \quad (3)$$

Salt, $Z(t)$

$$Z(t) = G(t) * V(t) = \text{Sum } Z_j(t) \quad (4)$$

Where

$$G(t) = \text{gm}(s)/L, \text{ salinity}$$

$$Z(t) = \text{gm}(s)*\text{af}/L$$

Also

$$Z'(t) = Z_{in}'(t) + Z_{ra}'(t) + Z_{sp}'(t) + Z_{ev}'(t) + Z_{se}'(t) + Z_{ex}'(t) \quad (5)$$

Where

$$Z_{in}'(t) = G_{in} * V_{in}'(t)$$

$$Z_{ra}'(t) = 0 = Z_{ev}'(t)$$

$$Z_{sp}'(t) = G_{sp} * V_{sp}'(t)$$

$$Z_{se}'(t) = -G(t) * V_{se}'(t)$$

$$Z_{ex}'(t) = -G(t) * V_{ex}'(t)$$

Then

$$Z'(t) = G_{in} * V_{in}'(t) + G_{sp} * V_{sp}'(t) - G(t) * [V_{se}'(t) + V_{ex}'(t)], \quad Z(0) = Z_o \quad (6)$$

Salinity, $G(t)$

$$G(t) = Z(t)/V(t), \text{ gm}(s)/L \quad (7)$$

Density, $R(t)$

$$R(t) = F(t), \text{ gm}(t)/L \quad (8)$$

%Salt, $S(t)$

$$S(t) = 100 * G(t)/R(t), \text{ gm}(s)/100 * \text{gm}(t) = \% \text{Salt} \quad (9)$$

Water, $W(t)$

$$W(t) = M(t) - Z(t), \text{ gm}(w)*\text{af}/L \quad (10a)$$

$$w(t) = R(t) - G(t), \text{ gm}(w)/L \quad (10b)$$

II.b Developing A Small Fresh Water Lake

Volume, $V(t)$

$$V'(t) = \{R_{in} \cdot V_{in}'(t) + R_o \cdot V_{ra}'(t) + R_{sp} \cdot V_{sp}'(t) - R_o \cdot V_{ev}'(t) - R'(t) \cdot V(t)\} / R(t) - V_{se}'(t) - V_{ex}'(t), \quad V(0) = V_o \quad (1)$$

Given Parameters, [11]

$$\begin{aligned} V_{ot} &= 2.7 \cdot 10^6 \text{ af} & \text{Elev} &= 6385 \text{ ft, surface} \\ A_{ot} &= 45,900 \text{ a} & R_o &= 1,000 \text{ gm(t)/L} \\ a_o &= b_o = 0.1 & R(t) &= R_o + E \cdot \exp(-g \cdot t) \\ V_o &= a_o \cdot V_{ot} = 0.27 \cdot 10^6 \text{ af} & E &= 62 \text{ gm(t)/L} \cdot y \\ A_o &= b_o \cdot A_{ot} = 4,590 \text{ a} & g &= 0.5362 \text{ 1/y} \\ H_{ra} &= 0.7 \text{ ft/y} & V_{in}'(t)_{ot} &= 152,496 \text{ af/y} \\ H_{ev} &= 3.75 \text{ ft/y} & V_{sp}'(t)_{ot} &= 48,000 \text{ af/y} \\ G_o &= 78 \text{ gm(s)/L} & V_{se}'(t)_{ot} &= 34,000 \text{ af/y} \end{aligned}$$

$$R'(t) = -g \cdot E \cdot \exp(-g \cdot t) \text{ gm(t)/L} \cdot y$$

If $V'(t) = 0$, $V(t) = V_o$, $A(V) = A_o$, $R_o \sim R_{in} \sim R_{sp}$, then

$$V_{ex}'(t) = [R_o/R(t)] \cdot [V_{in}'(t) + V_{ra}'(t) + V_{sp}'(t) - V_{ev}'(t) - R'(t) \cdot V(t)/R_o] - V_{se}'(t) \quad (2)$$

$$V(0) = V_o$$

Where

$$\begin{aligned} V_{in}'(t) &= V_{in}'(t)_{ot} = 152,496 \text{ af/y} = V_{in}' \\ V_{ra}'(t) &= b_o \cdot H_{ra} \cdot A_o = 0.1 \cdot 0.7 \cdot 45,900 = 3,213 \text{ af/y} = V_{ra}' \\ V_{sp}'(t) &= b_o \cdot V_{sp}'(t)_{ot} = 0.1 \cdot 48,000 = 4,800 \text{ af/y} = V_{sp}' \\ V_{ev}'(t) &= b_o \cdot H_{ev} \cdot A_o = 0.1 \cdot 3.75 \cdot 45,900 = 17,212.5 \text{ af/y} = V_{ev}' \\ V_{se}'(t) &= b_o \cdot V_{se}'(t)_{ot} = 0.1 \cdot 34,000 = 3,400 \text{ af/y} = V_{se}' \\ R'(t) \cdot V_o/R_o &= -g \cdot E \cdot \exp(-g \cdot t) \cdot V_o/R_o = -0.5362 \cdot 62 \cdot \exp(-G \cdot t)/1000 \\ &= -8,9875.9 \cdot E \cdot \exp(-g \cdot t) \text{ af/y} \end{aligned}$$

Thus

$$V_{ex}'(t) = [143,296.5 + 8975.9 \cdot \exp(-g \cdot t)] / [1.000 + 0.062 \cdot \exp(-g \cdot t)] - 3,400 \quad (3)$$

Where

$$\begin{aligned} V_{ex}'(0) &= 139,296.5 \text{ af/y} \\ V_{ex}'(\text{inf}) &= 139,896.5 \text{ af/y} \end{aligned}$$

Also

$$V_{ex}'(t) + V_{se}'(t) = [143,296 + 8,975.9 \cdot \exp(-g \cdot t)] / [1.000 + 0.062 \cdot \exp(-g \cdot t)] \quad (4)$$

Salt, $Z(t)$

$$Z(t) = G(t) \cdot V(t) = \text{Sum } Z_n(t) \quad (5a)$$

$$Z'(t) = Z_{in}'(t) + Z_{ra}'(t) + Z_{sp}'(t) + Z_{ev}'(t) + Z_{se}'(t) + Z_{ex}'(t) \quad (5b)$$

Given Parameters

$$\begin{aligned} G_{in} &= 0.064 \text{ gm(s)/L} & G_{sp} &= G_{in} \text{ gm(s)/L} \\ G_{se}(t) &= G(t) \text{ gm(s)/L} & G_{ex}(t) &= G(t) \text{ gm(s)/L} \\ G_{ra} &= 0 & G_{ev} &= 0 \\ G_o &= 78 \text{ gm(s)/L} & Z_o &= G_o \cdot V_o \end{aligned}$$

Now

$$\begin{aligned} Z_{in}'(t) &= G_{in} \cdot V_{in} = 0.064 \cdot 152496 = 8,759.9 \text{ gm(s)} \cdot \text{af/L} \cdot y \\ Z_{ra}'(t) &= 0 = Z_{ev}'(t) \\ Z_{sp}(t) &= G_{sp} \cdot V_{sp}' = 0.064 \cdot 4800 = 307.2 \text{ gm(s)} \cdot \text{af/L} \cdot y \\ Z_{se}'(t) &= -G(t) \cdot V_{se}'(t) = -Z(t) \cdot 3300/V(t) \\ &= -0.0121592 \cdot Z(t) \text{ gm(s)} \cdot \text{af/L} \cdot y \\ Z_{ex}'(t) &= -G(t) \cdot V_{ex}'(t) \text{ gm(s)} \cdot \text{af/L} \cdot y \end{aligned}$$

So

$$Z'(t) = G_{in} \cdot V_{in}' + G_{sp} \cdot V_{sp}' - G(t) \cdot [V_{se}'(t) + V_{ex}'(t)], \quad Z(0) = Z_o \quad (6)$$

IIb. Developing A Small Fresh Water Lake

Thus

$$Z'(t) = 10,066.9 + [0.530727 + 0.033244 \cdot \exp(-g \cdot t)] \cdot Z(t) / [1.000 + 0.062 \cdot \exp(-g \cdot t)]$$

$$Z(0) = Z_0 = 21.06 \cdot 10^6 \text{ gm(s)} \cdot \text{af/L} \quad (7)$$

A numerical solution to equation (7) is given in Table IV, in the following page.

Now

$$Z(t) = \text{gm(s)} \cdot \text{af/L} \quad (7a)$$

and

$$\text{gm(s)} \cdot \text{af/L} \cdot (L/m^3) \cdot (m^3/\text{af}) = \text{gm(s)} = (L/m^3) \cdot (m^3/\text{af}) \cdot Z(t) \quad (7b)$$

Thus

$$1.23347 \cdot 10^6 \cdot Z(t) = \text{gm(s)} \quad (7c)$$

Salinity, G(t)

$$G(t) = Z(t)/V(t) = Z(t)/V_0 = Z(t)/(0.27 \cdot 10^6) \text{ gm(s)/L} \quad (8)$$

So

$$G(t) = 78 \cdot \exp(-g \cdot t) + 0.0703 \cdot [1 - \exp(-g \cdot t)] \text{ gm(s)/L} \quad (9)$$

$$G(0) = 78 \text{ gm(s)/L}$$

$$G(\infty) = 0.0703 \text{ gm(s)/L}$$

$$g = 0.5362 \text{ 1/y}$$

Density, R(t)

$$R(t) = 1000 + 62 \cdot \exp(-g \cdot t) \text{ gm(t)/L} \quad (10)$$

$$R(0) = 1062 \text{ gm(t)/L}$$

$$R(\infty) = 1000 \text{ gm(t)/L}$$

%Salt, S(t)

$$S(t) = \text{gm(s)/L} \cdot L/\text{gm(t)} \cdot \text{gm(t)}/100\text{gm(t)} \text{ \%Salt} \quad (11a)$$

$$S(t) = 100 \cdot G(t)/R(t) \text{ \%Salt} \quad (11b)$$

$$S(t) = 100 \cdot [78 \cdot \exp(-g \cdot t) + 0.0703 \cdot (1 - \exp(-g \cdot t))] / [1.000 + 62 \cdot \exp(-g \cdot t)] \quad (11c)$$

$$S(0) = 7.345 \text{ \%Salt}$$

$$S(\infty) = 0.00703 \text{ \%Salt}$$

Water, W(t)

$$W(t) = M(t) - Z(t) = w(t) \cdot V(t) \text{ gm(w)} \cdot \text{af/L} \quad (12)$$

So

$$w(t) = R(t) - G(t) \text{ gm(w)/L} \quad (13)$$

$$w(t) = 999.9297 - 15.9297 \cdot \exp(-g \cdot t) \text{ gm(w)/L} \quad (14)$$

$$w(0) = 984 \text{ gm(w)/L}$$

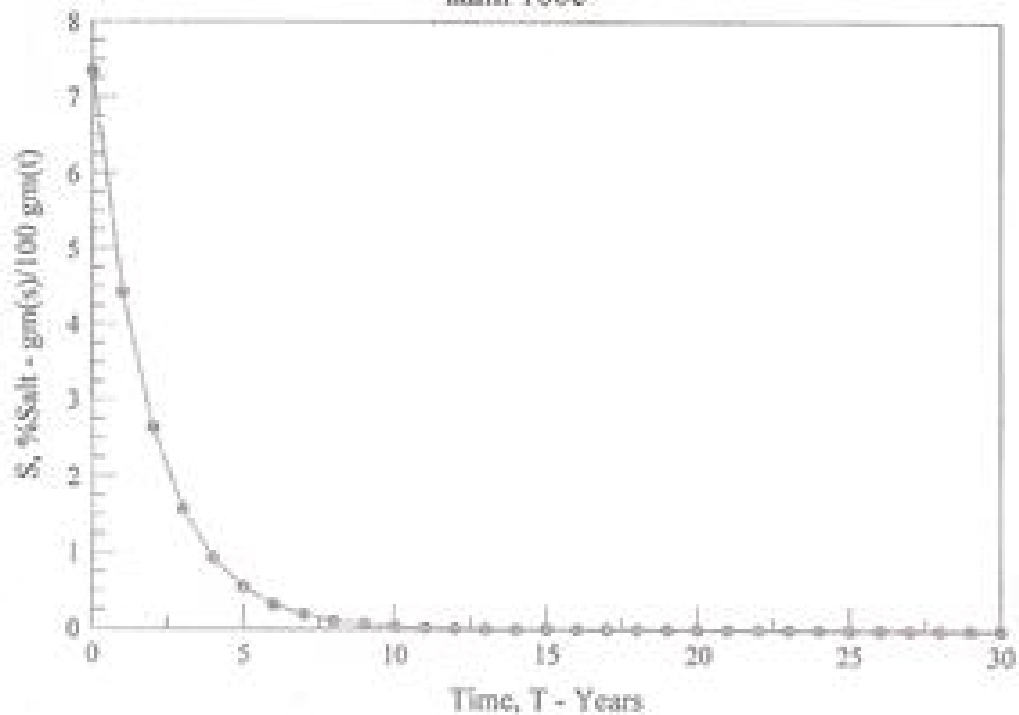
$$w(\infty) = 999.9297 \text{ gm(w)/L}$$

The solutions for G(t), R(t), S(t) and w(t) are also shown in Table IV.

Sheet: C:\Program Files\Microsoft Office\Excel\Book1.xls

	T, year	Q, km ³ /a	R, km ³ /a	S, % salt	m, km ³ /a	Total, km ³
1	0.000000	21.0000	20.0000	1000.0000	7.5000	990.0000
2	1.000000	11.0017	40.0000	1000.0000	4.4000	990.0000
3	2.000000	7.0042	57.0000	1000.0000	3.4000	990.0000
4	3.000000	4.0063	76.0000	1000.0000	2.6000	990.0000
5	4.000000	2.0088	9.0000	1000.0000	2.0000	990.0000
6	5.000000	1.0093	0.0000	1000.0000	1.0000	990.0000
7	6.000000	0.0094	0.0000	1000.0000	0.0000	990.0000
8	7.000000	0.0011	0.0000	1000.0000	0.0000	990.0000
9	8.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
10	9.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
11	10.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
12	11.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
13	12.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
14	13.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
15	14.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
16	15.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
17	16.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
18	17.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
19	18.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
20	19.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
21	20.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
22	21.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
23	22.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
24	23.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
25	24.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
26	25.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
27	26.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
28	27.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
29	28.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
30	29.000000	0.0000	0.0000	1000.0000	0.0000	990.0000
31	30.000000	0.0000	0.0000	1000.0000	0.0000	990.0000

Figure 1
Mono Lake
S, % Salt, vs Time
naml 100c



III. A Lake Complying With The Water Resources Board

From references [12] and [13],

Table V				
Year	Area, a	Volume, af	Salinity, gm(s)/L	Density, gm(t)/L
1999	45,900	$2.7 \cdot 10^6$	78	1062
2019	48,960	$3.0 \cdot 10^6$	69.3	1056
1963	48,739	$3.037 \cdot 10^6$	69	1056

Historically, as late as 1963, the Water Board requirements for the future were met in 1963 and for many previous years. [11] Using the conditions listed for the year 1999, in the equations for volume, salt, salinity, density and water content, as developed in Section IIb, and using the conditions listed in the parameters given below:

Parameters

$$\begin{aligned} V_{in}'(t) &= 144943 \text{ af/y} & V_{ra}'(t) &= H_{ra} \cdot A(V) \\ V_{sp}'(t) &= 48,000 \text{ af/y} & V_{ev}'(t) &= H_{ev} \cdot A(V) \\ V_{sp}'(t) &= 34,000 \cdot A(V)/A_o & V_{ex}'(t) &= 0 \\ H_{ra} &= 0.7 \text{ ft/y} & H_{ev} &= 3.75 \text{ ft/y} \end{aligned}$$

The results are close to the values listed for 2019.

CONCLUSION

The three possible futures for Mono Lake have been considered herein. It is seen that the Mono Lake PRO prospects are greater than the “sea water” PRO prospects. Additionally a small fresh water lake is possible in about 10 years. Satisfying the State Water Resources Board conditions was possible in the near past, and a possible future result is suggested.

Nomenclature

A = area, acre, a	af = acre-feet
ao = volume coefficient, dimensionless	Ao, Aot = starting areas, a
bo = area coefficient, dimensionless	Ch = water column factor, m/atm
Cw = work/(m ³ *atm)	E = density coefficient, gm(t)/L*atm
G = salinity, gm(s)/L	H = water column height, m
H _{ev} = evaporation factor, ft/y	H _{ra} = rain factor, ft/y
i = ions/molecule, dimensionless	L = liter
M = molarity, mol/L	M(t) = gm(t)*af/L
m = molality, mol/kg(t)	P = pressure, atm
[^] P = Osmotic pressure, atm	P _f = P fresh water, atm
P _s = P salt solution, atm	R = R(t), density, gm(t)/L
R' = Universal Gas Constant,	S = % salt, gm(s)/100gm(t)
t = time, y	V = volume acre-feet, af
V' = flow rate, af/y	V _o = initial volume, af
W = water mass, gm(w)*af/L	w = gm(w)/L
Z = salt, gm(s)*af/L	

Subscripts

in = in(flow)
 ra = rain
 sp = spring
 ev = evaporation
 se = seepage
 ex = out(flow)
 ot = original

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